2022

M.Sc.

2nd Semester Examination CHEMISTRY

PAPER—CEM-202

ORGANIC CHEMISTRY-II

Full Marks: 40

Time: 2 Hours

The figures in the margin indicate full marks.

Candidates are required to give their answers in their own words as far as practicable.

Illustrate the answers wherever necessary.

1. Predict the product(s) (any four, with plausible mechanism): 4×2

(a) (i)
$$\xrightarrow{\text{BF}_3}$$
 [A] (ii) $\xrightarrow{\text{D}}$ H $\xrightarrow{\text{LiNR}_2}$ [B]

(b) O
$$\longrightarrow$$
 O $\xrightarrow{1. \text{ LiN(SiMe}_3)_2, \text{ THF, } -30^{\circ}\text{C}}$ [C]

$$\frac{\text{H}_2/\text{Pd-C}}{\longrightarrow}$$
 [D]

(c)
$$\frac{1. \text{ NaH}}{2. \text{ C}_6 \text{H}_5 \text{SCl}} \text{ [E]} \xrightarrow{\text{heat}} \text{ [F]}$$

$$\frac{1. \text{ LDA, THF/-60°C}}{2. \text{ CH}_3 \text{I}} \rightarrow [\text{G}] \xrightarrow{\text{P(OCH}_3)_3} [\text{H}]$$

(d)
$$(\text{tBuCuCN})\text{Li} \longrightarrow [1] \xrightarrow{\text{HCl/H}_2\text{O}} [J]$$

(e)
$$\underbrace{ \begin{array}{c} H \\ O \\ \stackrel{=}{=} \end{array}}_{H} O \xrightarrow{1 \text{ LDA/THF/-78°C}} [K]$$

$$\frac{1. \text{ LDA/THF/-78°C}}{2. \text{ PhSeSePh}} \triangleright [L] \quad \frac{1. \text{ H}_2\text{O}_2/\text{THF/0°C}}{2. \text{ warm}} \triangleright [M]$$

(f)
$$\frac{\text{CH}_{3} \times \text{CP} = \text{O}}{\text{CH}_{3} \times \text{CO}_{2}\text{Cl}} = \text{O} \times \text{CH}_{3}\text{OH} = \text{O}_{2}\text{Cl} \times \text{O}_{$$

2. Answer any four questions:

- 4×4
- (a) What is Shapless Asymmetric Epoxidation? Give an example of it and explain with mechanism.
- (b) What are AD-mix-α and AD-mix-β? Show schematically the use of these reagents in organic synthesis. Write the structure of the product on reaction of AD-mix-β with Ethyl cinnamate.
 - (c) (i) What do you mean by convergent synthesis?
 - (ii) What is chelotropic reaction? Give an example.
- (d) (i) What do you mean by axial chirality? Give an example.
 - (ii) What do you mean by asymmetric induction?
- (e) Examine whether $[2\pi s + 2\pi s]$ cycloaddition reactions are thermally or photochemically allowed using correlation diagram.

(f) Give the retrosynthetic analysis as well as the forward synthesis for the following compound:

3. Answer any two questions:

2×8

(a) (i) Using retrosynthetic approach how will you synthesize the following compound:

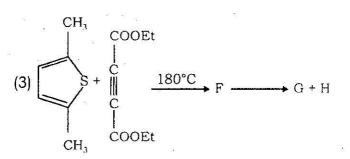
(ii) Predict the product(s) with plausible mechanism:

(1)
$$N \xrightarrow{M-CPBA} A$$

$$\downarrow \\ C_2H_5$$

 \longrightarrow B $\xrightarrow{\text{HgO}}$ C \longrightarrow D

(2)
$$H_2C = C - CH_3 \xrightarrow{1. Pr_2'NLi/THF} E$$
Ph 3. H



(b) Predict the product(s) (any four, with plausible mechanism):

(i)
$$\frac{\text{OCH}_3}{\text{Na/liq. NH}_3}$$

$$\frac{\text{Na/liq. NH}_3}{\text{H}_3\text{O}^{\dagger} \text{ (mild)}}$$

- (c) (i) Draw all the diasteroisomers of 1, 2, 3-trimethyl cyclohexane and comment on their relative stabilities.
 - (ii) What do you mean by 3-alkyl ketone effect? Explain with an example.

(iii) Predict the product(s) with proper justification:

Ph — C — C — OH + CH₃Li —
$$CH_3$$

3+3+2

(d) Synthesize the following from suitable starting materials:

(any two from i - iv and any one from v - vi):

(iv)
$$H_3C$$
 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

$$\begin{array}{cccc} & & & & & & \\ & & & & & \\ \text{(v)} & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & \\ & & \\ & \\ & & \\ &$$

2×2