2019

Major 3rd Semester Examination

INDUSTRIAL CHEMISTRY

Paper - C 6-T

Full Marks: 40 Time: 2 Hours

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words as far as practicable.

1.	Answer	any five	questions	:	2×5=10

- (a) What is the basic principle of paper chromatography?
- (b) What information you get from the Retardation factor value?
- (c) What is a chromatogram?
- (d) In thin layer chromatography mention the nature of stationary phase and mobile phase. 2

(e)	What are the common adsorbents used in TLC '			
		2		
(f)	What is a flame ionization detector?	2		
(g)	Write down the principles of HPLC.	2		
(h)	n) Mention major components of a high performan			
	liquid chromatograph.	2		
Answer any four questions from the below: 5×4=20				

(a) (i) Write down the principle of ion-exchange chromatograpy.

2.

(ii) Mention few commonly used ion-exchangers.

(b) (i) Mention the solvent delivery system and solvent injection system in HPLC. 2

(ii) Mention four applications of ion-exchange chromatography. 3

(c) (i) Write down the functions of solvents used in column chromatography.

Mention few solvents used in column chromatography. 5

(d) (i) What is the role of detectors in

	chromatography? 2			
	(ii) What are the factors that affect column efficiency in chromatography?			
	(e) With a clear sketch dicuss Radial paper chromatography.			
	(f) Mention different steps in chromatographic separation.			
3.	Answer any one question from the below: $10 \times 1 = 1$			
	(a) With respect to gas chromatography answer the following:			
	(i) Principle 2			
	(ii) Mobile phase and stationary phase 2			
	(iii) Schematic diagram of a capillary column gas chromatography, with special mention of the major components.			
	(iv) Solute detection system.			
	(b) (i) What are the limitations of pape			

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(ii) Write down the principle of GC-MS combination in analysing complex mixture. Give a schematic diagram of the same. 5

(iii) Mention basic operations involved in TLC.

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