

**2018**

**M.Sc.**

**1st Semester Examination**

**CHEMISTRY**

**PAPER—CEM-102**

*Full Marks : 40*

*Time : 2 Hours*

*The figures in the margin indicate full marks.*

*Candidates are required to give their answers in their own words as far as practicable.*

*Illustrate the answers wherever necessary.*

**( Organic Chemistry )**

**Group—A**

Answer any four questions : 4×2

1. What is Olefin Metathesis reaction? What is Grubbs catalyst?
2. What is 'biomimetic control' in chemical transformation?

*(Turn Over)*

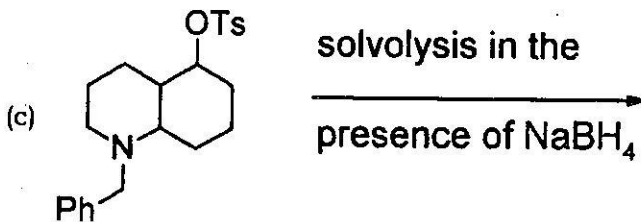
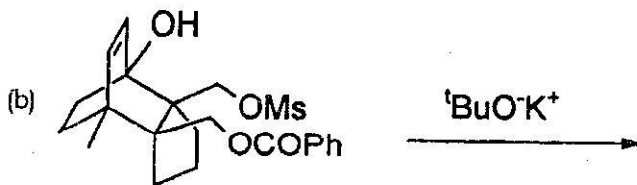
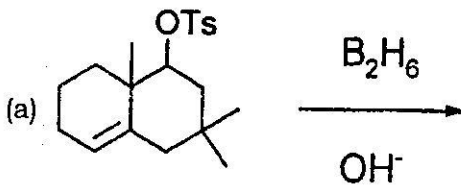
3. What is "biogenetic isoprene rule" ?
4. What is Barton reaction ? Explain schematically with mechanism.
5. What is multicomponent reaction ? Write its significance.
6. What is phase transfer catalyst ? Give in example and explain its mechanism.
7. What is Grob Fragmentation ?
8. Plant based chemicals can be termed as *Renewable Chemicals*. Explain.

**Group—B**

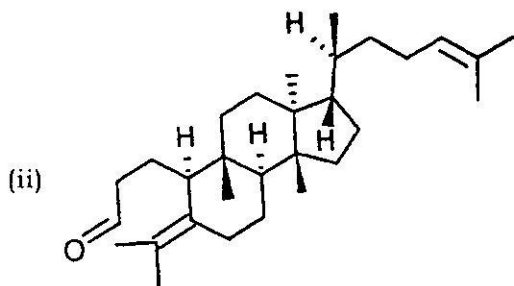
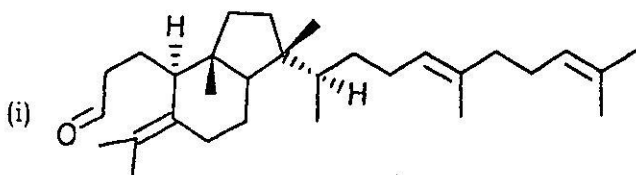
Answer any *four* questions : 4×4

9. Explain the formation of the following from squalene epoxide by applying the "biogenetic isoprene rule" with at least three examples each (answer any two) :
  - (i) monocyclic triterpenoids
  - (ii) bicyclic triterpenoids
  - (iii) tricyclic triterpenoids

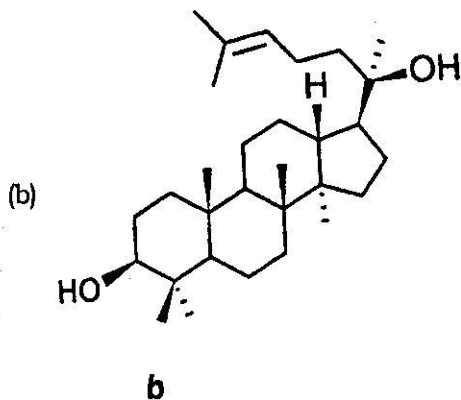
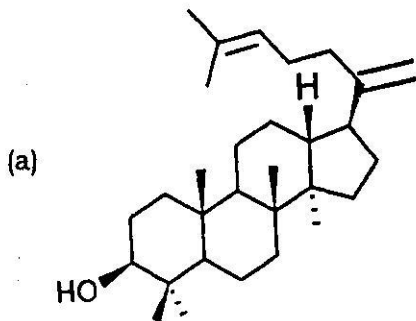
10. Predict the products (with plausible mechanism) (any two) :

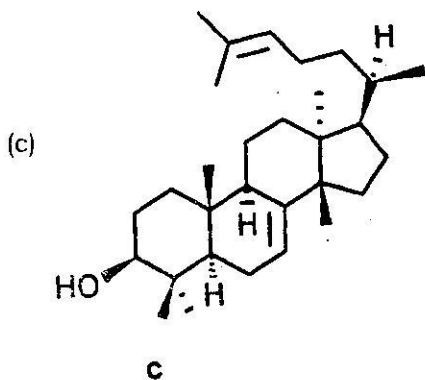


11. Synthesize the following from squalene by applying biogenetic isoprene rule and Grob fragmentation :

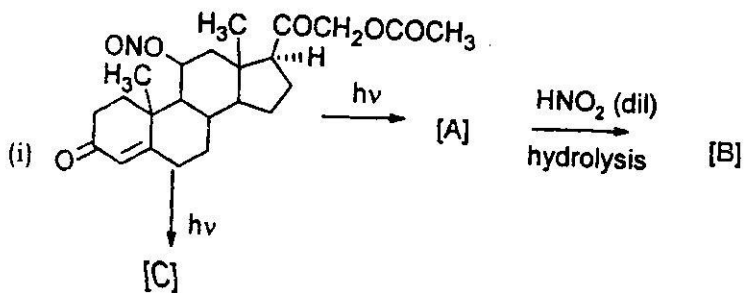


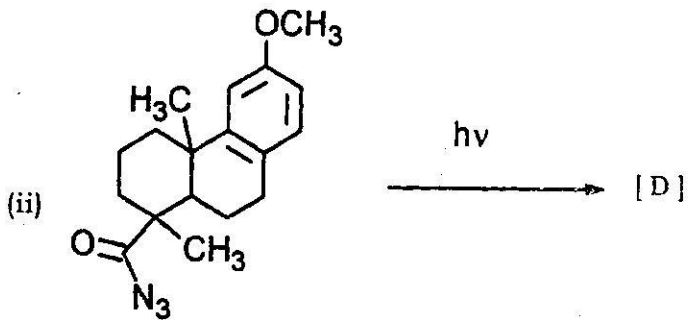
12. Synthesize the following 6-6-6-5 tetracyclic triterpenoids from squalene by applying biogenetic isoprene rule (answer any two) :



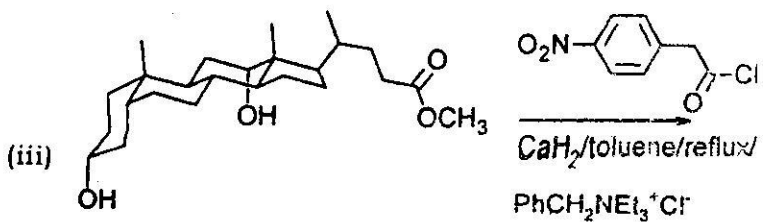
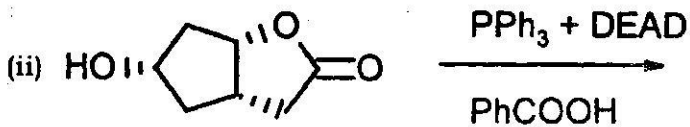
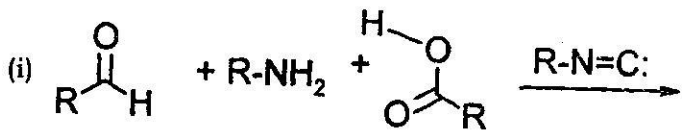


13. Carry out the following transformations with plausible mechanism.

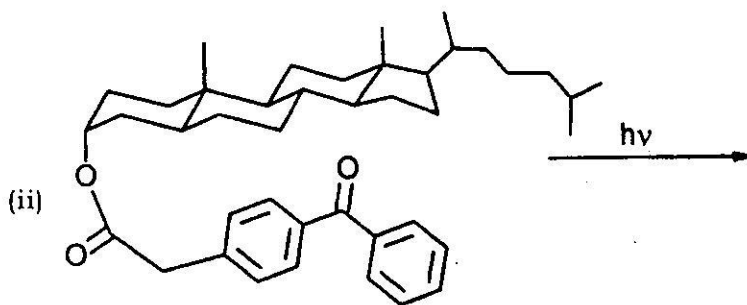
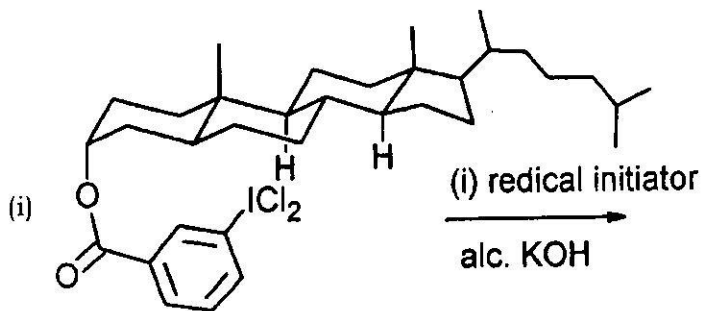




14. Predict the products with plausible mechanism (any two) :

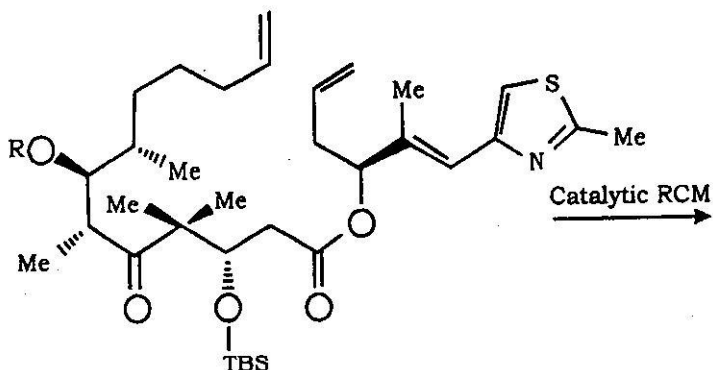


15. Predict the products with plausible mechanism :





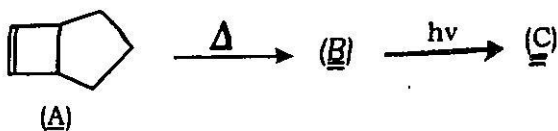
16. What is ring closing metathesis (RCM) reaction? Predict the product with plausible mechanism.



**Group—C**

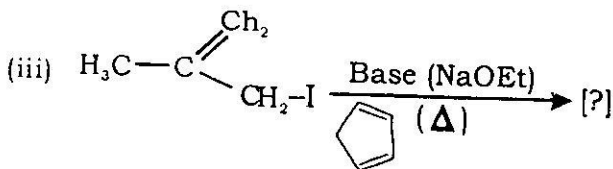
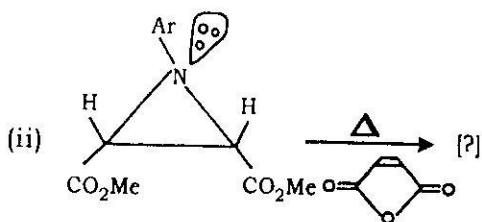
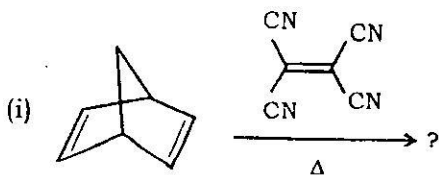
Answer any two questions : 2×8

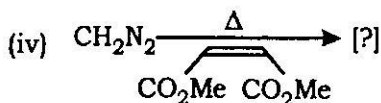
17. (a) The following compound (A) under thermal ( $\Delta$ ) reaction to yield compound (B), which is contrary to Woodward-Hofman Rule, but compound (B) on photochemical irradiation ( $h\nu$ ) produces compound (C) where the reverse reaction is not followed, as shown : 2×2+2×2



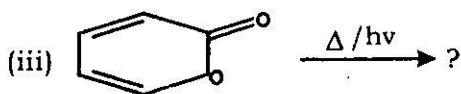
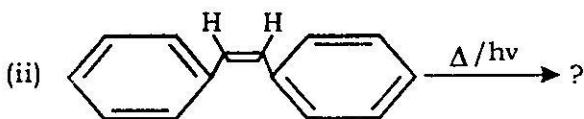
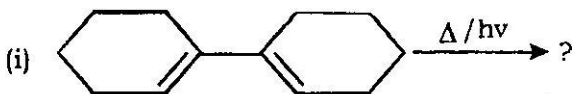
Identify the compound (B) and (C) explain the reason behind this observation with justification.

- (b) Predict the products of the following reactions indication F.O.I. (any two)

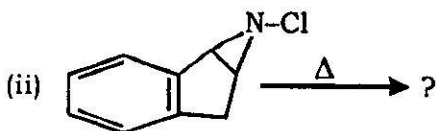
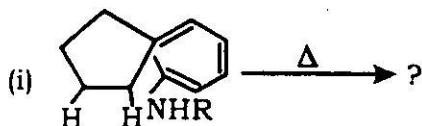




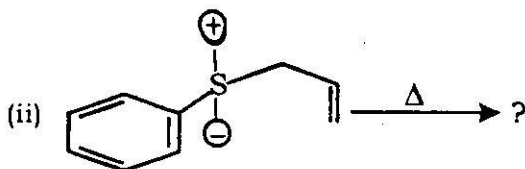
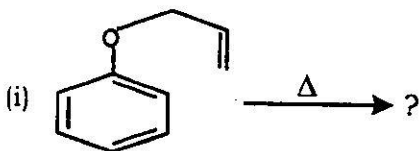
18. (a) The following compounds under both thermal ( $\Delta$ ) and photochemical ( $h\nu$ ) reaction but the reaction process through only one path to produce the preferred product. Explain the formation of the product in each case and rationalise indicating Frontier-Orbital interactions (F.O.I.) (any two):

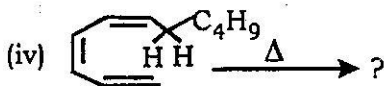
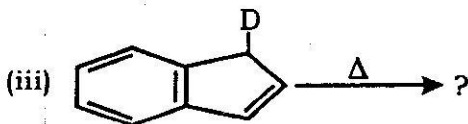


- (b) Predict the product/s of the following reaction indicating F.O.I. (any one) : 2×3+2



19. What is (i,g) sigmatropic shift? Illustrate with examples and hence predict the product of the following reactions (attempt any three) : 2+3×2



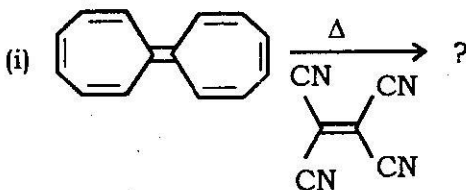


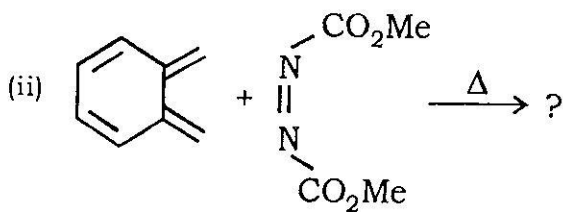
20. (a) Draw the correlation diagram of the following interconversion under thermal condition : 4+2×2



And indicate the symmetry allowed path under this condition.

- (b) Predict the product/s of the following reaction with *F.O.I.* (attempt any two) : 2×4





(iii)

