OLD

Part-II 3-Tier

2016

CHEMISTRY

(General)

PAPER-III

(PRACTICAL)

Full Marks: 100

Time: 6 Hours

The figures in the right-hand margin indicate full marks.

Group-A

- 1. Make a systematic qualitative analysis of the organic compound supplied in the container marked "O".
 - (i) Give the name and procedure of the experiment and detect the presence or absence of the special elements namely nitrogen (N), sulphur (S) and halogens (Cl, Br, I) in the supplied compound.

- (ii) Detect the presence or absence of the following functional groups:
 - COOH, Phenolic OH, Ketonic >C = O,
 - CHO (Aldehyde), $-NH_2$ (Aromatic) and
 - -NO₂ (Aromatic).
- (iii) Give the confirmative test of any one of the functional groups present in the supplied compound.
- (iv) Naming of the functional group(s) detected. 2

Group-B

- 2. Carry out any one of the following two experiments (Experiment should be selected by lottery):
 - (a) Prepare 250 ml. approx. $\left(\frac{N}{10}\right)$ standard solution of

potassium dichromate by accurate weighing in an analytical balance. Mention the type of balance used for weighing.

Find out the amount of Fe⁺⁺ in gms / litre in Mohr's salt solution supplied in the bottle marked "V":

- (i) Weighing in chemical balance.
- (ii) Volumetric results upto four decimal places.

25

		tect any three radicals (acid/base/interfering the supplied inorganic sample marked "I".	acid)
	(Igr	nore ${\rm CO_3}^{2-}$ radical present in the sample.)	
	(i)	Physical properties.	3
	(ii)	Dry tests for acid and basic radicals (only postests).	sitive 6
	(iii)	Dry tests for interfering acid radicals.	6
	(iv)	Wet tests for acid and basic radicals (only postests).	sitive 12
	(v)	Confirmative tests for three radicals.	3
	(vi)	Correct naming of three radicals in ionic form	. 3
		Group—C	
3.	Laboratory Note Book.		12
4.	Internal performance / Attendance.		10
5.	Viva-Voce.		